

**using specific heat answer pdf**

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Worksheet- Calculations involving Specific Heat 1. For  $q = m c \Delta T$  : identify each variables by name & the units associated with it.  $q$  = amount of heat (J)  $m$  = mass (grams)  $c$  = specific heat (J/g $^{\circ}$ C)  $\Delta T$  = change in temperature ( $^{\circ}$ C)

**Worksheet- Calculations involving Specific Heat**

Specific Heat Worksheet Name (in ink):  $C = q/m\Delta T$ , where  $q$  = heat energy,  $m$  = mass, and  $T$  = temperature Remember,  $\Delta T = (T_{\text{final}} - T_{\text{initial}})$ . Show all work and proper units. Answers are provided at the end of the worksheet without units. 1. A 15.75-g piece of iron sorbs 1086.75 joules of heat energy, and its

**Specific Heat Wksht20130116145212867**

Print Answer Key PDF Take Now Schedule Copy. Print Answer Key (Only the test content will print) Specific Heat Answer Key. 1. ... The specific heats of gases are generally expressed as molar specific heats because the number of moles is a more convenient measure of an amount of gas.

**Specific Heat Answer Key - HelpTeaching.com**

Activity $\hat{e}$ Specific Heat Capacity Handout Answer Key 2 4. To heat the hot chocolate to the optimal temperature of 57  $^{\circ}$ C, how much energy is needed?  $Q = mc\Delta T$   $Q = (50 \text{ g})(3.9 \text{ J/g } ^{\circ}\text{C})(57 ^{\circ}\text{C} - 40 ^{\circ}\text{C})$   $Q = 3,315 \text{ J}$  Analysis Questions Answers will vary, depending on collected data. 1. Water has a specific heat of 4.18 J/g  $^{\circ}$ C.

**Specific Heat Capacity Handout Answer Key**

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Specific Heat Problems 1) How much heat must be absorbed by 375 grams of water to raise its temperature by 25 $^{\circ}$  C? 2) What mass of water can be heated from 25.0 $^{\circ}$  C to 50.0 $^{\circ}$  C by the addition of 2825 J? 3) What is the final temperature when 625 grams of water at 75.0 $^{\circ}$  C loses 7.96 x 10<sup>4</sup> J?

**Specific Heat Problems - mmsphyschem.com**

Name:\_\_\_\_\_ Per:\_\_\_\_\_ Worksheet- Introduction to Specific Heat Capacities Heating substances in the sun: The following table shows the temperature after 10.0 g of 4 different substances have been in direct sunlight for up to 60 minutes.

**Name: Per: Worksheet- Introduction to Specific Heat Capacities**

Heat Capacity\_Calorimetry Worksheet Answers - Download as Word Doc (.doc Since specific heat can be described using the equation  $q = m \times c_p \times \Delta T$ , or  $c_p = q$  energy would have less of an effect on a material

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