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Ammonium sulfate, $(\text{NH}_4)_2\text{SO}_4$, is often used for salting out because of its high solubility, which allows for solutions of very high ionic strength, low price, and availability of pure material. Additionally, NH_4^+ and SO_4^{2-} are at the ends of their respective Hofmeister series and have been shown to stabilize protein structure (Burgess, 2009). Some proteins follow a reversal of the ...

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Ammonium sulfate precipitation - Wikipedia

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Point spread function - Wikipedia

Curriculum Vitae Page 3 Dr. Lawrence J. DeLucas Honors Recognized as one of the scientists who could shape the 21st century in an article published by "The Sunday Times" of London titled "The Brains Behind the 21st Century" (January, 1999)

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